

! " 0.03%

! (H <sub>3</sub> PO <sub>4</sub> )	" 85.0%	" 85.0%	85.5%
# \$ % & ' ( )	* 25	* 25	+ 25
, - . ( / H+O ) % mmol / g	* 0.0002	* 0.0002	+ 0.0002
1 2 3 (Cl)	* 0.0003%	* 0.0005%	+ 0.0003%
4. 5 (SO <sub>4</sub> )	* 0.003%	* 0.01%	+ 0.003%
6. 5 (NO <sub>3</sub> )	* 0.0005%	* 0.0005%	+ 0.0005%
7 (As)	* 0.00005%	* 0.0001%	+ 0.00005%
8 (Fe)	* 0.002%	* 0.002%	+ 0.002%
9 (Na)	* 0.05%	* 0.06%	+ 0.05%
: (K)	* 0.005%	* 0.006%	+ 0.005%
; (Mn)	* 0.0002%	* 0.0005%	+ 0.0002%
< (Ni)	* 0.0005%	* 0.001%	+ 0.0005%
= (Cu)	* 0.0005%	* 0.001%	+ 0.0005%
> (Zn)	* 0.001%	* 0.001%	+ 0.001%
? (Cd)	* 0.0005%	* 0.0005%	+ 0.0005%
@ (Pb)	* 0.001%	* 0.001%	+ 0.001%
AB3C ( / H <sub>3</sub> PO <sub>3</sub> O )	* 0.01%	* 0.03%	+ 0.01%