

! (H <sub>3</sub> PO <sub>4</sub> )	" 85.0%	" 85.0%	85.6%
# \$%&' ( )	* 25	* 25	+ 25
, - . (/ H+O)%mmol/g	* 0.0002	* 0.0002	+ 0.0002
1 2 3(Cl)	* 0.0003%	* 0.0005%	+ 0.0005%
4. 5(SO <sub>4</sub> )	* 0.003%	* 0.01%	+ 0.01%
6. 5(NO <sub>3</sub> )	* 0.0005%	* 0.0005%	+ 0.0005%
7 (As)	* 0.00005%	* 0.0001%	+ 0.0001%
( Al )	* 0.0005%	0.25001%	+ 1%

01