

! (H PO ₄ ³⁻)	" 85.0%	" 85.0%	85.4%
# \$%&' ()	* 25	* 25	+ 25
, - . (/ H+O)%mmol /g	* 0.0002	* 0.0002	+ 0.0002
1 2 3(Cl)	* 0.0003%	* 0.0005%	+ 0.0005%
4. 5(SO ₄ ²⁻)	* 0.003%	* 0.01%	+ 0.01%
6. 5(NO ₃ ⁻)	* 0.0005%	* 0.0005%	+ 0.0005%
7(As)	* 0.00005%	* 0.0001%	+ 0.0001%
(Al)	* 0.0005%	0.25001%	+ 0.0005%