

! (H P0↔)	" 85.0%	" 85.0%	85.6%
# \$%&' ( )	* 25	* 25	+ 25
, - . (/ H+O)%mmol /g	* 0.0002	* 0.0002	+ 0.0002
1 2 3(Cl)	* 0.0003%	*	+ 0.0003%