

! (H ₂ PO ₄ ⁻)	" 85.0%	" 85.0%	85.5%
# \$%&' ()	* 25	* 25	+ 25
, - . (/ H ₂ O)%mmol/g	* 0.0002	* 0.0002	+ 0.0002
1 2 3 (Cl)	* 0.0003%	* 0.0005%	+ 0.0005%
4. 5 (SO ₄ ²⁻)	* 0.003%	* 0.01%	+ 0.01%
6. 5 (NO ₃ ⁻)	* 0.0005%	* 0.0005%	+ 0.0005%
7 (As)	* 0.00005%	* 0.0001%	+ 0.0001%
8 (Fe)	* 0.002%	* 0.002%	+ 0.002%
9 (Na)	* 0.05%	* 0.06%	+ 0.06%
: (K)	* 0.005%	* 0.006%	+ 0.006%
; (Mn)	* 0.0002%	* 0.0005%	+ 0.0005%
< (Ni)	* 0.0005%	* 0.001%	+ 0.001%
= (Cu)	* 0.0005%	* 0.001%	+ 0.001%
> (Zn)	* 0.001%	* 0.001%	+ 0.001%
? (Cd)	* 0.0005%	* 0.0005%	+ 0.0005%
@ (Pb)	* 0.001%	* 0.001%	+ 0.001%

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